

# Early Events Induced by the Elicitor Cryptogein in Tobacco Cells: Involvement of a Plasma Membrane NADPH Oxidase and Activation of Glycolysis and the Pentose Phosphate Pathway

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Application of the elicitor cryptogein to tobacco (cv Xanthi) is known to evoke external medium alkalinization, active oxygen species production, and phytoalexin synthesis. These are all dependent on an influx of calcium. We show here that cryptogein also induces calcium-dependent plasma membrane depolarization, chloride efflux, cytoplasm acidification, and NADPH oxidation without changes in NAD<sup>+</sup> and ATP levels, indicating that the elicitor-activated redox system, responsible for active oxygen species production, uses NADPH *in vivo*. NADPH oxidation activates the functioning of the pentose phosphate pathway, leading to a decrease in glucose 6-phosphate and to the accumulation of glyceraldehyde 3-phosphate, 3- and 2-phosphoglyceric acid, and phosphoenolpyruvate. By inhibiting the pentose phosphate pathway, we demonstrate that the activation of the plasma membrane NADPH oxidase is responsible for active oxygen species production, external alkalinization, and acidification of the cytoplasm. A model is proposed for the organization of the cryptogein responses measured to date.

## INTRODUCTION

The interaction of *Phytophthora cryptogea* with tobacco cultivar Xanthi results in a hypersensitive reaction of the plant triggered by the proteinaceous elicitor cryptogein produced by the fungus (Billard et al., 1988; Ricci et al., 1989). Cryptogein elicits hypersensitive-like necroses on tobacco leaves and also protects the plant against various pathogens, especially *P. parasitica* var *nicotianae*.

For the past few years, we have been studying the mode of action of this elicitor by using tobacco cultivar Xanthi cell suspensions. We previously reported different effects of cryptogein on tobacco cells, including capsidiol and ethylene production (Milat et al., 1991), alkalinization of the extracellular medium and K<sup>+</sup> efflux (Blein et al., 1991), active oxygen species (AOS) production (Bottin et al., 1994), protein phosphorylation (Viard et al., 1994), changes in lipid composition (Tavernier et al., 1995a), calcium influx (Tavernier et al., 1995b), and early changes in gene expression (Suty et al., 1995; Petitot et al., 1997). Our results have led us to conclude that protein phosphorylation followed by calcium influx

may be involved in the initial steps of cryptogein signal transduction (Tavernier et al., 1995b). Moreover, we have characterized specific high-affinity binding sites for cryptogein ( $K_d$  of 2 nM) in tobacco leaf plasma membranes (Wendehenne et al., 1995) and in plasma membranes from suspension-cultured tobacco cells (S. Bourque, A. Lebrun-Garcia, and A. Pugin, unpublished data).

The transient and fast production of AOS, including superoxide, H<sub>2</sub>O<sub>2</sub>, and the hydroxyl radical, by using cell suspensions treated with various elicitors is now well documented (Apostol et al., 1989; Sutherland, 1991; Bottin et al., 1994; Chen and Heath, 1994; Mehdy, 1994; Auh and Murphy, 1995; Tavernier et al., 1995b; Mathieu et al., 1996a). Whereas the antimicrobial activity of AOS has been recognized in animals, it is still not clear whether in plants, the AOS produced in response to pathogens or elicitors may have direct antimicrobial effects (Baker and Orlandi, 1995). They are assumed to be involved in other defense reactions, such as lipid peroxidation (Keppler and Novacky, 1987; Adam et al., 1989; Vera-Estrella et al., 1993; Stallaert et al., 1995), oxidative cross-linking of cell wall proteins (Bradley et al., 1992; Brisson et al., 1994), peroxidative polymerization of

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cinnamoyl alcohols leading to the lignification of cell walls (Hammerschmidt and Kuc, 1982; Kohle et al., 1984), and phytoalexin production (Doke, 1983; Apostol et al., 1989; Devlin and Gustine, 1992; Vera-Estrella et al., 1993). However, other results have shown that  $\text{H}_2\text{O}_2$  is not a primary signal for phytoalexin production (Davis et al., 1993; Levine et al., 1994). Moreover, it has been reported that  $\text{H}_2\text{O}_2$  from the oxidative burst triggers the hypersensitive death of challenged cells and functions in surrounding cells as a diffusible signal for induction of defense genes encoding enzymes involved in cellular protection (Levine et al., 1994; Tenhaken et al., 1995) or associated with systemic acquired resistance (Chen et al., 1993).

Many results indicate that  $\text{O}_2^{\cdot-}$  synthesis occurs at the extracellular surface of the plasma membrane via a one-electron reduction of molecular oxygen (Doke, 1985; Auh and Murphy, 1995). For instance, Doke (1985) reported NADPH-dependent  $\text{O}_2^{\cdot-}$  production by using microsomes from potato tubers infected by pathogen; in mammalian neutrophils, the production of superoxide  $\text{O}_2^{\cdot-}$  as an antimicrobial agent involves a multicomponent NADPH oxidase that is assembled and functions at the plasma membrane (Babior, 1992; Segal and Abo, 1993). As previously reported (Rubinstein and Stern, 1986; Guern et al., 1988; Marré et al., 1988; Grabov and Böttger, 1994), the transfer of electrons from a cytosolic electron donor (NAD[P]H) to electron acceptors at the extracellular surface of the plasma membrane may lead to (1) an acidification of the cytosol due to the release of protons from the reducing agent; (2) an alkalization of the extracellular medium due to the consumption of protons by different reactions (superoxide/hydroperoxide equilibrium [ $\text{O}_2^{\cdot-} + \text{H}^+ \rightarrow \text{HO}_2\cdot$ ] and the dismutation of  $\text{O}_2^{\cdot-}$  to  $\text{H}_2\text{O}_2$  and  $\text{O}_2$  either spontaneously or catalyzed with superoxide dismutases [ $\text{HO}_2\cdot + \text{O}_2^{\cdot-} + \text{H}^+ \rightarrow \text{H}_2\text{O}_2 + \text{O}_2$ ;  $2\text{O}_2^{\cdot-} + 2\text{H}^+ \rightarrow \text{H}_2\text{O}_2 + \text{O}_2$ ; Baker and Orlandi, 1995]); and (3) a decrease of the plasma membrane potential due to the electron transfer at the extracellular surface.

Thus, the activation of a plant plasma membrane redox system by pathogens or elicitors (Arnott and Murphy, 1991; Nürnberger et al., 1994; Tavernier et al., 1995b) may be responsible for cytosol acidification, previously reported in elicited cells (Ojalvo et al., 1987; Mathieu et al., 1991, 1996b), and plasma membrane depolarization, which is a common feature of elicited cells (Mathieu et al., 1991; Kuchitsu et al., 1993; Thain et al., 1995). These effects could participate in signal transduction and amplification.

Evidence supports the similarity of NAD(P)H oxidase in plant cells and NADPH oxidase in phagocytic animal cells involved in  $\text{H}_2\text{O}_2$  production. In particular, antibodies raised against components of the neutrophil NADPH oxidase, responsible for the generation of  $\text{O}_2^{\cdot-}$  during phagocytosis, cross-react with polypeptides of similar molecular masses in plants (Levine et al., 1994; Kieffer et al., 1997; Xing et al., 1997). Moreover, plant NAD(P)H oxidases are inhibited (Levine et al., 1994; Nürnberger et al., 1994; Dwyer et al., 1996) by diphenylene iodonium (DPI), a well-known inhibitor of the

mammalian neutrophil oxidase (Cross and Jones, 1986). Nevertheless, no direct evidence indicates that NADPH is the substrate for this plant oxidase. In contrast, it was suggested that  $\text{H}_2\text{O}_2$  could originate from other oxidases (Bolwell et al., 1995), because in many plant-elicitor interactions, the NADPH-to-NADP ratio remained constant during elicitation (Fahrendorf et al., 1995; Carpita et al., 1996).

By using phosphorus-31 nuclear magnetic resonance ( $^{31}\text{P}$ -NMR) to measure NADPH and NADP<sup>+</sup> levels and cytoplasmic pH changes, electrophysiological measurements to monitor plasma membrane potential, and selective electrodes to measure chloride efflux in cryptogeiin-treated tobacco cells, we show that cryptogeiin activates a plasma membrane NADPH oxidase whose activity depends on calcium influx and that is responsible for cytoplasm acidification but not for plasma membrane depolarization. NADPH oxidation activates the cytosolic pentose phosphate pathway, with the carbohydrate products being converted into glycolytic compounds that accumulate.

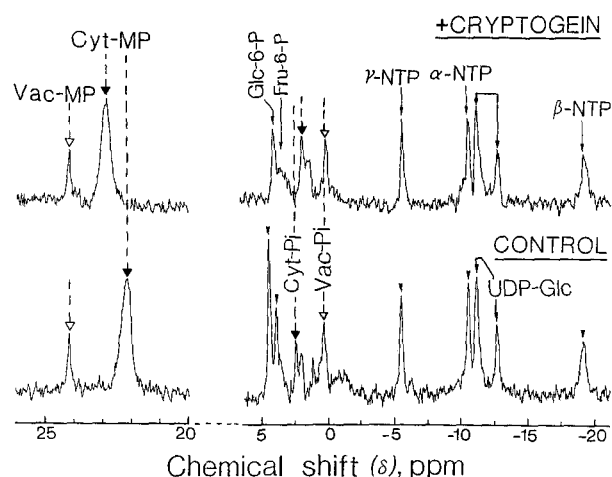
## RESULTS

### Cryptogeiin-Induced Cytoplasm Acidification and Calcium Dependence of the Response

$^{31}\text{P}$ -NMR is a well-suited method for determining both changes in the pool size of endogenous phosphate compounds and intracellular pHs (Roberts and Jardetzky, 1981; Martin et al., 1982; Roby et al., 1987; Guern et al., 1989; Aubert et al., 1994).

Typical  $^{31}\text{P}$ -NMR spectra of control and cryptogeiin-treated cells are illustrated in Figure 1. The peaks were assigned to specific compounds whose localization was deduced from the appropriate controls, taking into account the pH of the surrounding medium (Martin et al., 1982). In particular, two peaks of Pi were resolved at  $\sim 2.25$  and  $0.25$  ppm corresponding to the cytoplasmic Pi and to the vacuolar Pi at pH 7.5 and pH 5.5, respectively. Other peaks were identified as cytoplasmic compounds: glucose 6-phosphate (Glc-6-P), fructose 6-phosphate (Fru-6-P), nucleoside triphosphates, and UDP-glucose (UDP-Glc). In addition, methylphosphonate, which accumulated inside cells from an external supply (see Methods), was also resolved in two peaks corresponding to vacuolar and cytoplasmic pools.

Ten minutes after the addition of 100 nM cryptogeiin in the circulating medium, the chemical resonance of the peak corresponding to the cytosolic Pi shifted from 2.25 to 2.0 ppm (Figure 1), indicating a  $0.4 \pm 0.1$  ( $n = 3$ ) pH unit acidification of the cytoplasm. No vacuolar pH change occurred during this period. The intensity of cytoplasmic acidification was confirmed and made more precise by using the change of the chemical shift of cytoplasmic methylphosphonate from 22.2 to 22.9 ppm, which corresponds to cytoplasmic pH of 7.5 and 7.05, respectively (pH drop of  $0.45 \pm 0.05$ ;  $n = 3$ ). On the other hand, cryptogeiin (100 to 500 nM) did not



**Figure 1.** Representative Proton-Decoupled in Vivo  $^{31}\text{P}$ -NMR Spectra (161.92 MHz) of Control and Cryptogein-Treated Tobacco Cells.

The cells (10 g wet weight) were packed in a 25-mm NMR tube, as described in Methods, and continuously perfused at a flow rate of 50 mL per min with a well-oxygenated manganese-free culture medium maintained at pH 6.5 and  $20^\circ\text{C}$ , as described by Roby et al. (1987). In the windows of analysis, the cell volume comprised  $\sim 50\%$  of the total (cell plus perfusing medium) volume. The total perfusing medium was 500 mL. Spectra were recorded by using a multinuclear probe under the conditions given in Methods. Each spectrum is the result of 1000 scans cumulated over 10 min. The spectrum corresponding to cryptogein-treated cells was obtained 10 min after the addition of 100 nM cryptogein. Data are representative of three independent experiments. Peak assignments are as follows: Glc-6-P, glucose 6-P; Fru-6-P, fructose 6-P; Cyt-Pi, cytoplasmic phosphate; Vac-Pi, vacuolar phosphate; NTP, nucleoside triphosphate; UDP-Glc, uridine 5'-diphosphate- $\alpha$ -D-glucose; Cyt-MP, cytoplasmic methylphosphonate; and Vac-MP, vacuolar methylphosphonate.

induce any cytoplasmic acidification in *Acer pseudoplatanus* cells (data not shown), indicating that, as with the other effects of cryptogein, cytoplasmic acidification was specific to the tobacco cell-cryptogein interaction.

In the presence of 100  $\mu\text{M}$  lanthanum, which is known to block the entrance of calcium into tobacco cells treated with cryptogein (Tavernier et al., 1995b), cryptogein did not induce any cytoplasmic acidification during a 2-hr treatment (data not shown), indicating that cytoplasmic acidification depends on the entrance of calcium.

#### Activation of an NADPH Oxidase

The spectra of cells treated with cryptogein for 10 min (Figure 1) also revealed a significant decrease in Glc-6-P concentration from 5 to 3 mM. This decrease in Glc-6-P could indicate activation of glycolysis and the tricarboxylic acid

cycle (corresponding to an increased consumption of NADH) or of the pentose phosphate pathway (corresponding to an increased consumption of NADPH) or both.

$^{31}\text{P}$ -NMR analysis of perchloric acid extracts prepared from 30-min-treated cells (at the end of the oxidative burst) and from control cells revealed that the NADPH-to-NADP $^{+}$  ratio, estimated from the upfield peaks of NADP $^{+}$  and NADPH located at 3.61 and 3.63 plus 3.64 ppm, respectively, strongly decreased in treated cells (Figure 2). Cytoplasmic NADPH levels decreased from 0.3 mM in control cells to 0.05 mM in treated cells, whereas the NAD $^{+}$  level (0.42 mM) did not change (see Table 1 and Methods for identifications and calculations). After perchloric acid extraction, NADH gave an upfield quadruplet, which overlapped with UDP-Glc and  $\alpha$ -ATP peaks, like that of NADPH. Because NADH gives no upfield peak, it was not measurable.

Moreover, spectra analysis indicated a large accumulation in treated cells of products belonging to the second section of the glycolysis: glyceraldehyde 3-phosphate, 3-phosphoglyceric acid, 2-phosphoglyceric acid, and phosphoenolpyruvate, whereas the ATP levels did not change (Figure 2 and Table 1).

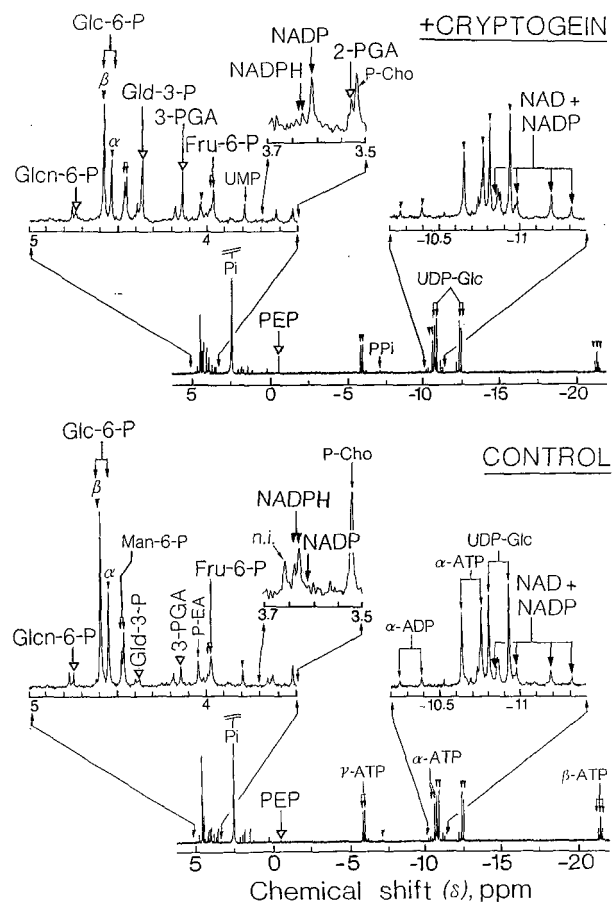
A similar decrease in NADPH level and a similar accumulation of glycolytic products were also measured in 15-min-treated cells during the oxidative burst. After a 1-hr treatment, cells partially recovered their initial status (data not shown).

These results clearly indicate that cryptogein induces a large oxidation of NADPH that leads to activation of the pentose phosphate pathway, whose first products are NADPH and ribulose 5-phosphate. NADPH is immediately oxidized by the plasma membrane oxidase, and ribulose 5-phosphate is converted into glyceraldehyde 3-phosphate and Fru-6-P. Both of these compounds are involved in glycolysis and/or glucose synthesis (Figure 3).

#### Origin of Cytoplasmic Acidification

To determine whether redox system activation was involved in cytosolic acidification and extracellular medium alkalinization, as suggested above, and to verify that NADPH was the electron donor, we inhibited the pentose phosphate pathway in two independent ways and monitored the consequences of these inhibitions on three effects of cryptogein, namely,  $\text{H}_2\text{O}_2$  production, cytosolic acidification, and extracellular alkalinization.

The first enzyme of the pentose phosphate pathway is Glc-6-P dehydrogenase, which transforms Glc-6-P into 6-phosphogluconate. Glucosamine 6-phosphate (N-Glc-6-P) is a well-known competitive inhibitor of the enzyme (Glaser and Brown, 1955). To introduce the inhibitor into tobacco cells, 5 mM glucosamine was added for 1 hr to cell suspensions in place of glucose. Under this condition, cells took up glucosamine that was rapidly phosphorylated to N-Glc-6-P,



**Figure 2.** Representative Proton-Decoupled in Vitro  $^{31}\text{P}$ -NMR Spectra (Perchloric Extracts, Expanded Scales from 5 to 3.5 ppm and from  $-10.2$  to  $-11.4$  ppm, between Arrows) of Control Tobacco Cells and Cells Incubated in the Presence of 50 nM Cryptogein for 30 Min.

Perchloric extracts were prepared from 9 g wet weight of cells, as described in Methods. The samples (3 mL) containing 300  $\mu\text{L}$  of  $\text{D}_2\text{O}$  were analyzed for 8 hr (8192 scans) at  $20^\circ\text{C}$  with a 10-mm multinuclear probe tuned at 161.92 MHz. Data are representative of three independent experiments. Because of perchloric acid extraction, NADPH is transformed in a way giving upfield quadruplets, which are masked by UDP-Glc and  $\alpha$ -ATP (see also Roberts et al., 1997). The downfield resonance of NADPH is split into two peaks at 3.63 and 3.64 ppm.  $\text{NADP}^+$ , which is not transformed by perchloric acid extraction, gives a downfield peak at 3.61 ppm and an upfield quadruplet at  $-10.87$ ,  $-10.99$ ,  $-11.19$ , and  $-11.31$  ppm. The increase of the upfield quadruplet signal, which is identical to the increase of the downfield peaks, corresponds to  $\text{NADP}^+$  production when, after addition of cryptogein, NADPH is oxidized in  $\text{NADP}^+$ .  $\text{NAD}^+$ , which is also stable after perchloric acid treatment and gives an upfield quadruplet superimposing exactly to that of  $\text{NADP}^+$ , remained constant. Peak assignments are as follows: Glc-6-P, glucose 6-P; Glc-3-P, gluconate 6-P; Gld-3-P, glyceraldehyde 3-P; 3-PGA, 3-phosphoglyceric acid; 2-PGA, 2-phosphoglyceric acid; PEP, phosphoenolpyruvate; UDP-Glc, uridine 5'-diphosphate- $\alpha$ -D-glucose; UMP, uridine monophosphate; Fru-6-P, fructose 6-P; Man-6-P, mannose 6-P;

as determined by  $^{31}\text{P}$ -NMR spectra of perchloric acid extracts of treated cells (Figure 4). The large decrease in 6-phosphogluconate level (Figure 4) agrees with the expected inhibition of Glc-6-P dehydrogenase. Moreover, the  $\text{NADPH}$ -to- $\text{NADP}^+$  ratio estimated from perchloric acid extracts decreased strongly in glucosamine-treated cells (Table 2). In glucosamine-treated cells, cryptogein still induced calcium influx (Figure 5A) but did not induce any  $\text{H}_2\text{O}_2$  production (Figure 5B), cytosolic acidification (data not shown), or extracellular alkalization (Figure 5C).

These results favor an NADPH oxidase activated by the elicitor and responsible for AOS production, cytoplasmic acidification, and extracellular medium alkalization. However, inhibiting the effects of cryptogein after treatment with glucosamine could be interpreted alternatively as an inhibition of elicitation by glucosamine itself, as previously reported with other monosaccharides (Basse et al., 1992, 1993), or by AOS scavenging. These alternative interpretations may be rejected because in the presence of 5 mM glucose, glucosamine was not taken up by the cells, and 100 nM cryptogein induced all of the previously described responses (data not shown).

In addition, to confirm the involvement of an NADPH oxidase in cryptogein-induced AOS production, cytoplasmic acidification, and extracellular medium alkalization, experiments were performed using tobacco cells whose pentose phosphate pathways were inhibited in another way. It was previously reported that the addition of glycerol in place of sucrose in the culture medium of *A. pseudoplatanus* cells resulted in nonfunctioning of the cytosolic and plastidial pentose phosphate pathways (Aubert et al., 1994).

Thus, when tobacco cells were cultivated for 2 weeks in the presence of 50 mM glycerol in place of sucrose, the pentose phosphate pathway was not supplied with its initial substrate Glc-6-P. At that moment, the NMR analysis of perchloric acid extracts revealed that the  $\text{NADPH}$ -to- $\text{NADP}^+$  ratio was largely decreased (Table 2). Under these conditions, 100 nM cryptogein did not induce AOS production, cytosolic acidification, or extracellular medium alkalization.

Moreover, when tobacco cells were treated with 25  $\mu\text{M}$  DPI, an inhibitor of the mammalian neutrophil NADPH oxidase, 100 nM cryptogein did not induce AOS production and cytoplasmic acidification (data not shown). Taken together, these results indicate that the activation of the plasma membrane redox system responsible for  $\text{H}_2\text{O}_2$  production was involved in cytosolic and extracellular pH modifications and that this redox system oxidized NADPH in vivo.

P-EA, phosphorylethanolamine; P-Cho, phosphorylcholine; n.i., not identified. Open and solid arrowheads show intermediates of the glycolytic and pentose phosphate pathways and pyridine nucleotides, respectively.

**Table 1.** Modifications of the Levels of Phosphorylated Compounds in Tobacco Cells Treated for 30 Min with 50 nM Cryptogein Measured Using  $^{31}\text{P}$ -NMR Analysis<sup>a</sup>

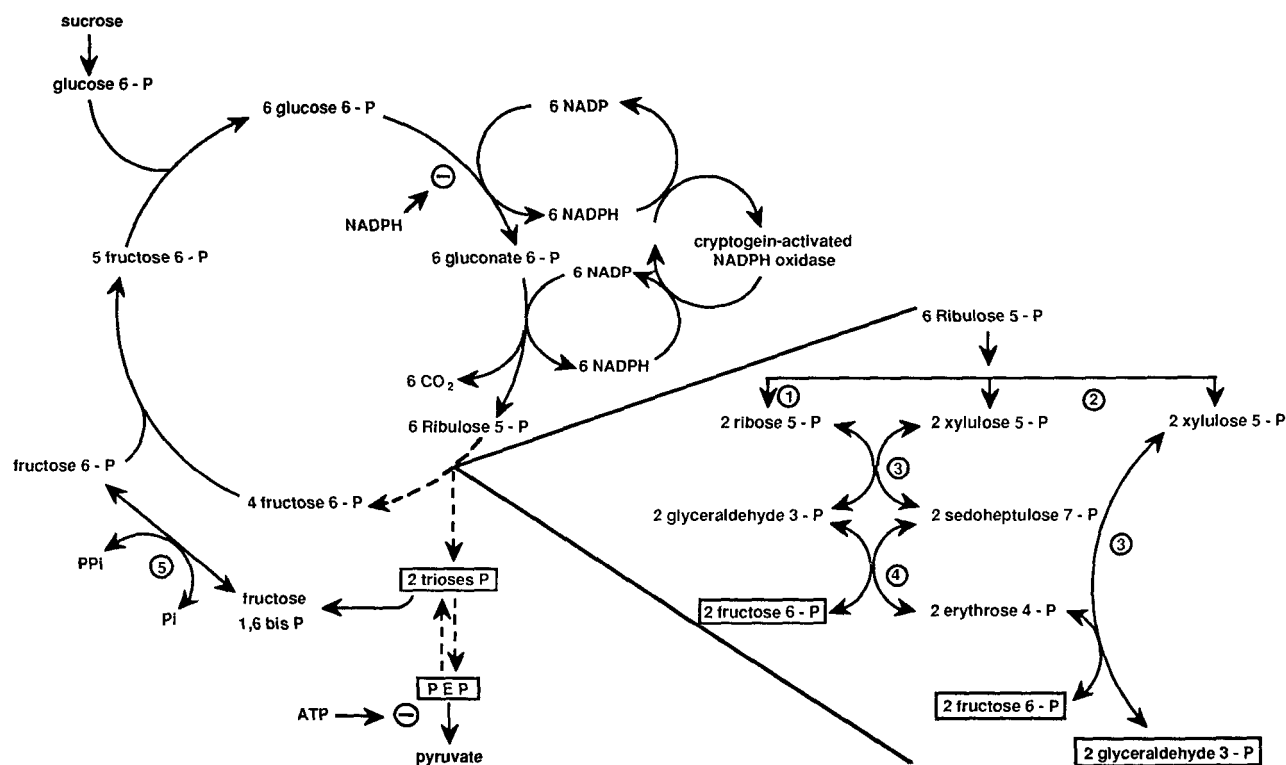
Compounds	Glc-6-P	Fru-6-P	Man-6-P	Glc-6-P	Glc-3-P	3-PGA	2-PGA	PEP	ATP	UDP-Glc	NAD <sup>+</sup>	NADP <sup>+</sup>	NADPH
Control	5.0	1.2	1.7	0.3	0.1	0.4	<0.05	0.2	2.2	3.4	0.42	<0.05	0.3
Cryptogein	3.0	1.3	1.2	0.15	2.2	1.0	0.15	0.9	2.0	3.4	0.42	0.3	<0.05

<sup>a</sup>Data are calculated from perchloric acid extracts, as indicated in Methods, and expressed as millimolar concentrations in the cytoplasm. Data are from a representative experiment (three assays). Abbreviations are as given in the legends to Figures 1, 2, and 4.

### Plasma Membrane Depolarization, Chloride Efflux, and Calcium Dependence

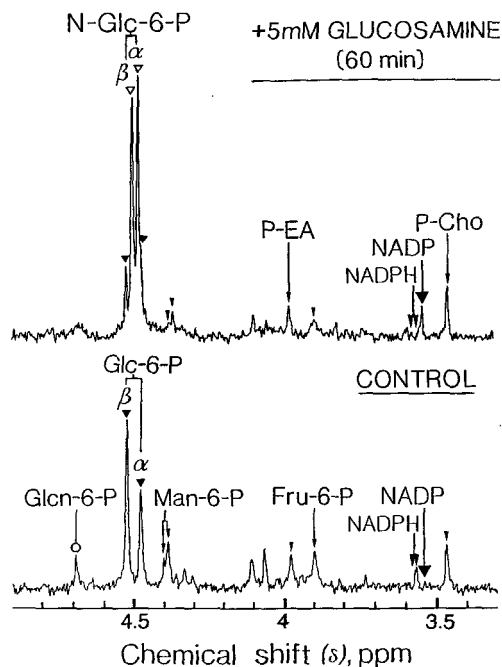
The plasma membrane potential ( $E_m$ ) of control tobacco cells was  $-153.8 \pm 14.9$  mV ( $n = 5$ ). Figure 6A shows that

the addition of 150 nM cryptogein depolarized the membrane to  $-35.6 \pm 21.2$  mV ( $n = 5$ ). The whole depolarization occurred within 1 min, after a lag period of  $4.3 \pm 1.9$  min. The addition of 250  $\mu$ M lanthanum to the perfusion medium induced the repolarization of the membrane.



**Figure 3.** Schematic Representation of the Cytosolic Oxidative Pentose Phosphate Pathway Activated by Cryptogein to Supply the Demand for NADPH for  $O_2^-$  Production at the Plasma Membrane.

The oxidative reactions of the pathway catalyzed by Glc-6-P dehydrogenase and phosphogluconate dehydrogenase both generate NADPH. These reactions are essentially irreversible. The reversible section of this pathway is involved in the regeneration of hexose phosphate and triose phosphates from ribulose 5-phosphate. The pentose phosphate pathway can bring about the complete breakdown of Glc-6-P to CO<sub>2</sub> if Fru-6-P and glyceraldehyde 3-phosphate (triose phosphate) are completely recycled. Recycling of triose phosphates requires the presence of a cytosolic pyrophosphate-linked phosphofructokinase. The most effective point of control of this pathway is at the initial step. Indeed, NADPH is a potent inhibitor of Glc-6-P dehydrogenase (competitive with respect to NADP<sup>+</sup>). Finally, this scheme also explains the marked increase observed in the concentration of cytoplasmic glyceraldehyde 3-phosphate, 3-phosphoglyceric acid, and phosphoenolpyruvate during the rapid re-oxidation of NADPH at the plasma membrane. Indeed, most of the reactions catalyzed by glyceraldehyde 3-phosphate dehydrogenase, phosphoglyceric acid kinase, phosphoglyceromutase, and enolase are close to equilibrium. Circled numbers indicate involved enzymes: 1, ribose-phosphate isomerase; 2, ribulose-phosphate epimerase; 3, transketolase; 4, transaldolase; 5, PPI-linked phosphofructokinase. Dashed arrows indicate that intermediate compounds are not given.



**Figure 4.** Representative Proton-Decoupled in Vitro  $^{31}\text{P}$ -NMR Spectra (Perchloric Extracts, Expanded Scale from 4.8 to 3.4 ppm) of Control Tobacco Cells and Cells Incubated in the Presence of 5 mM Glucosamine for 60 Min.

Perchloric extracts were prepared from 9 g wet weight of cells, as described in Methods. The samples (3 mL) containing 300  $\mu\text{L}$  of  $\text{D}_2\text{O}$  were analyzed for 1 hr (1024 scans) at  $20^\circ\text{C}$  with a 10-mm multinuclear probe tuned at 161.92 MHz, as detailed in Methods. Data are representative of three independent experiments. Peak assignments are as follows: Glcn-6-P, gluconate 6-P; Glc-6-P, glucose 6-P; Fru-6-P, fructose 6-P; Man-6-P, mannose 6-P; N-Glc-6-P, glucosamine 6-P; P-EA, phosphorylethanolamine; P-Chol, phosphorylcholine.  $\text{NADP}^+$  (NADP) and NADPH peaks correspond to the 2'-P of this oxidized/reduced pyridine nucleotide. Open symbols indicate outstanding compounds in this assay.

In the same way, the addition of 250  $\mu\text{M}$  lanthanum in the perfusion medium before cryptogein treatment prevented membrane depolarization (Figure 6B). Washing lanthanum from the medium while retaining cryptogein restored the cryptogein-induced depolarization. A second addition of lanthanum repolarized the membrane.

To verify whether the plasma membrane depolarization resulted from the activation of the plasma membrane redox system, we measured plasma membrane potential variation induced by cryptogein by using cells previously perfused for 30 min in the presence of 5 mM glucosamine. Under these conditions, in which the plasma membrane redox system was inhibited, 150 nM cryptogein induced the previously described depolarization (Figure 7A). Here, too, this depolarization was reversed by 250  $\mu\text{M}$  lanthanum. Thus, the plasma membrane depolarization induced by cryptogein did not result from the plasma membrane redox system activation.

This result was confirmed by treating tobacco cells with 25  $\mu\text{M}$  DPI. Under these conditions, in which 150 nM cryptogein did not activate  $\text{H}_2\text{O}_2$  production (data not shown), cryptogein induced a plasma membrane depolarization reversed by lanthanum (Figure 7B).

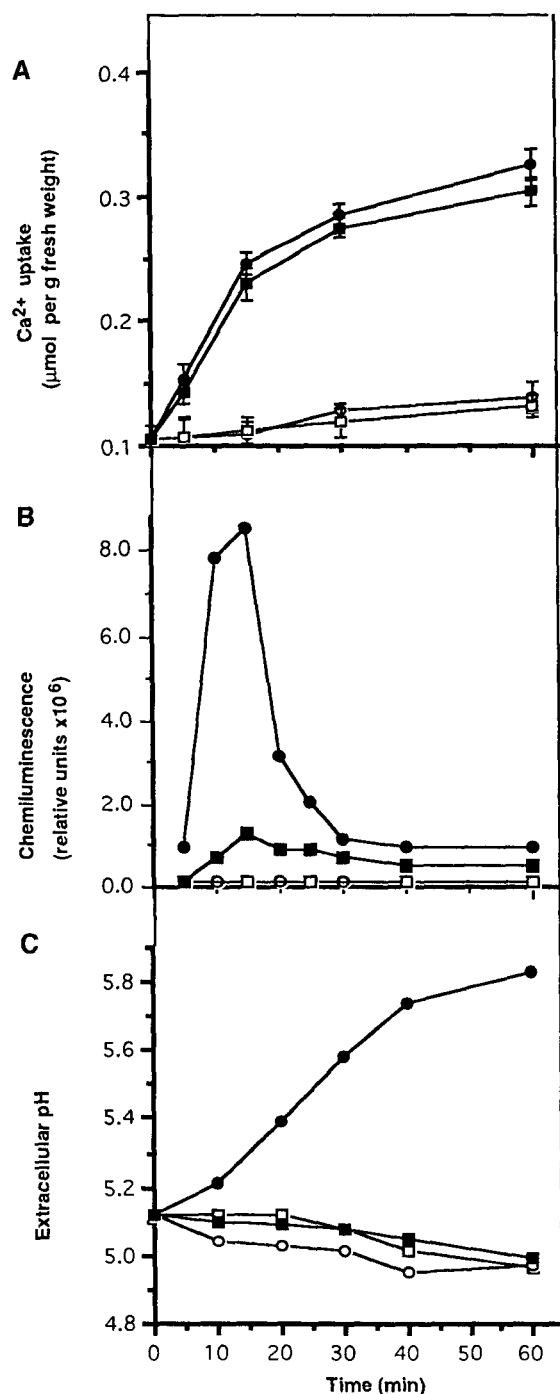
Because anion efflux could be involved in cryptogein-induced membrane depolarization, the influence of cryptogein on the exchanges of chloride was analyzed. Figure 8 shows that treating tobacco cells with 50 nM cryptogein induced a significant efflux of  $\text{Cl}^-$  (30 nmol of  $\text{Cl}^-$  per min per g fresh weight of cells). This efflux was strongly reduced (by  $\sim 90\%$ ) in the presence of 250  $\mu\text{M}$  lanthanum, which blocks calcium entry.

None of these effects may be attributed to lethal effects of cryptogein. Indeed, using fluorescein diacetate and neutral red, we did not observe any dead cells during the first 3 hr of treatment with 100 nM cryptogein.

## DISCUSSION

In this study, we provide evidence that the cryptogein-activated redox system, which is responsible for AOS production, oxidized NADPH in vivo. The activation of this plasma membrane NADPH oxidase is also responsible for the elicitor-induced cytoplasmic acidification and external medium alkalization, at least during the initial phase of these responses. In contrast, cryptogein-induced membrane depolarization appears to be independent of the activation of this redox system. In addition, we observed that cryptogein did not induce a cytoplasmic pH change, extracellular medium alkalization, or AOS production in *A. pseudoplatanus* cell suspensions, confirming the specificity of the tobacco-cryptogein interaction.

The large decrease in the NADPH-to- $\text{NADP}^+$  ratio in treated cells without significant changes in  $\text{NAD}^+$  and ATP levels on the one hand and the large accumulation of glycolytic products on the other (Figure 2 and Table 1) clearly indicate that NADPH is the cofactor of the plasma membrane oxidase activated by cryptogein. The oxidation of NADPH by the plasma membrane redox system leads to the activation of the pentose phosphate pathway, which essentially is used for furnishing NADPH, with ribulose 5-phosphate being recycled in Fru-6-P and glyceraldehyde 3-phosphate. Fru-6-P is probably isomerized mainly in Glc-6-P that feeds the pentose phosphate pathway again, whereas glyceraldehyde 3-phosphate is converted in glycolytic products that accumulate (Figure 3). The large increase in glyceraldehyde 3-phosphate (from 0.1 to 2.2 mM), 2- and 3-phosphoglyceric acid, and phosphoenolpyruvate suggests that the activity of pyruvate kinase did not allow the further metabolism of the excess of phosphoenolpyruvate to respiration substrate. Indeed, the concentration of ATP, a potent inhibitor of pyruvate kinase, remained constant and high during cryptogein treatment.



**Figure 5.** Calcium Influx, AOS Production, and Extracellular Alkalinization Induced by Cryptogein in Tobacco Cell Suspensions in the Presence or Absence of Glucosamine.

(A) Calcium influx.

(B) AOS production.

(C) Extracellular alkalinization.

(A) to (C) show induction by 50 nM cryptogein (dark symbols) in tobacco cell suspensions previously incubated for 1 hr in the presence

**Table 2.** Modifications of NADPH and NADP<sup>+</sup> Levels Induced in Tobacco Cells by Glucosamine Treatment of Glycerol Feeding and as Measured Using <sup>31</sup>P-NMR<sup>a</sup>

Treatment <sup>a</sup>	Control Cells	Glucosamine Treatment	Glycerol-Fed Cells
NADP <sup>+</sup>	0.1 ± 0.05	0.30 ± 0.05	0.20 ± 0.08
NADPH	0.4 ± 0.1	ND <sup>b</sup> (<0.005)	ND <sup>b</sup> (<0.005)

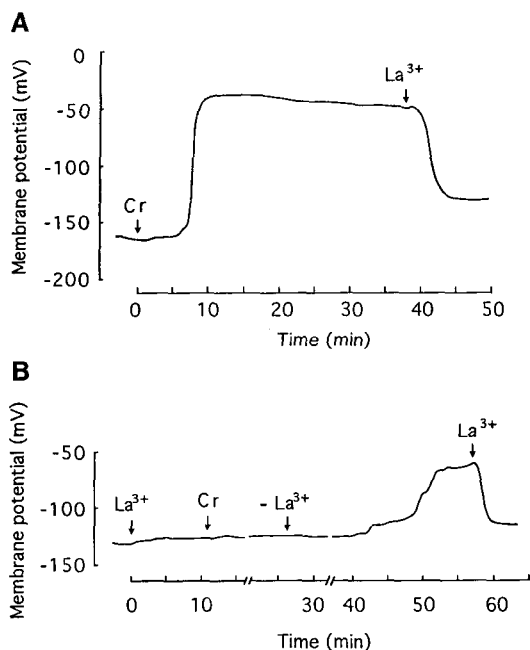
<sup>a</sup>NADP<sup>+</sup> and NADPH were measured by <sup>31</sup>P-NMR from perchloric acid extracts, as indicated in Methods, and expressed as millimolar concentrations in the cytoplasmic compartment. The standard deviation is given for 90% from three different experiments.

<sup>b</sup>ND, not detected (<0.005 mM).

In contrast, the recycling of triose phosphates back to fructose 1,6-bisphosphate and further to Fru-6-P (from the back reaction of PPI-linked phosphofructokinase) was favored. Indeed, a rapid substrate cycle between the triose phosphates and the hexose phosphates was already observed in the cytosol of the heterotrophic *Chenopodium rubrum* cells by Hatzfeld and Stitt (1990). They also demonstrated that this cycle is due to the activity of the PPI-linked phosphofructokinase. Fructose 1,6-phosphatase activity is much too low to catalyze this back reaction. The magnitude of this back reaction would automatically respond to the needs of the oxidative pentose phosphate pathway. Thus, in contrast to Glc-6-P, Fru-6-P remained nearly constant (Table 1). Taken together, these results show that elicitation triggers a flux of carbohydrates from the pentose phosphate pathway to the lower and upper sections of the glycolysis with an active recycling toward Glc-6-P (Figure 3).

To provide further evidence concerning the oxidase co-factor in vivo and to evaluate the role played by the activation of the plasma membrane redox system in the induction of cytosolic acidification, we inhibited the pentose phosphate pathway in vivo or suppressed the supply of Glc-6-P to this pathway. The decrease in 6-phosphogluconate normally produced by Glc-6-P dehydrogenase and the decrease in the NADPH-to-NADP<sup>+</sup> ratio (Figure 4 and Table 2) confirmed the effective inhibition of the pentose phosphate pathway. Under these conditions in which calcium influx induced by cryptogein was not affected, three characteristic responses to the elicitor, namely, cytoplasmic acidification, external medium alkalinization, and AOS production, were

(black squares) or absence (black circles) of 5 mM glucosamine. Control cells are with (open squares) or without (open circles) glucosamine. The data are representative of five experiments.



**Figure 6.** Tobacco Cell Plasma Membrane Depolarization Induced by Cryptogein and Its Inhibition by Lanthanum.

**(A)** The addition of 150 nM cryptogein (Cr; first arrow) depolarized the membrane to  $-40$  mV after a lag period of 7 min. Treatment with 250  $\mu$ M lanthanum ( $\text{La}^{3+}$ ; second arrow) induced the repolarization of the membrane. Cells were equilibrated and continuously perfused during voltage measurements with a minimal medium (10 mM Mes, 5 mM sucrose, 0.5 mM  $\text{CaSO}_4$ , and 3 mM KOH, pH 5.7). Data are representative of four experiments.

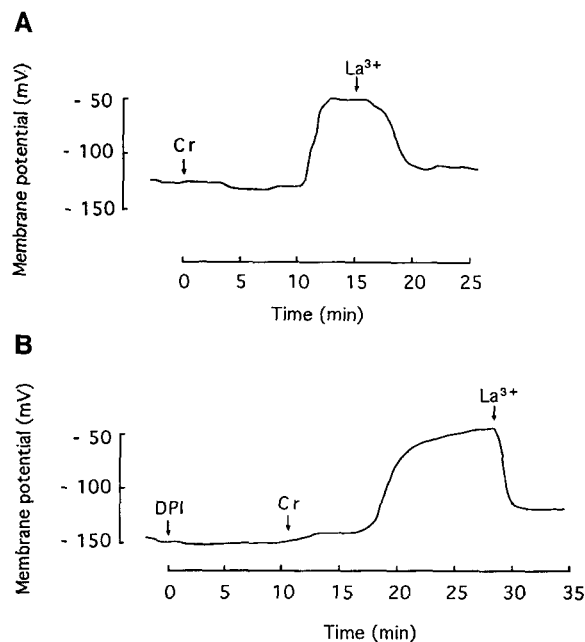
**(B)** Lanthanum (250  $\mu$ M) added to the perfusion medium (first arrow) prevented membrane depolarization by 150 nM cryptogein (second arrow). Washing lanthanum from the medium (third arrow) while keeping cryptogein in the perfusing medium restored the cryptogein-induced depolarization. A second addition of lanthanum (fourth arrow) repolarized the membrane. The same bath medium as given in **(A)** was used. Data are representative of four experiments.

abolished, indicating the involvement of the NADPH oxidase in pH modifications.

However, the time courses of pH changes and AOS production did not correlate, that is, alkalinization peaked long after the burst was over (Figure 5) and cytoplasmic acidification remained constant from 10 min to  $>1$  hr. This suggests that additional mechanisms involved in pH regulation are mobilized in a second step to stabilize and/or amplify the pH changes due to the activity of the plasma membrane oxidase. For example, inhibition of the plasma membrane  $\text{H}^+$ -ATPase, by an excess of cytoplasmic calcium and/or after oxidation of thiol groups by AOS, could explain long-lasting pH changes. This assumption rests on the observation that

ATP levels did not decrease in cryptogein-treated cells (Figure 1 and Table 1), whereas the large and sustained plasma membrane depolarization and cytoplasmic acidification should induce an activation of this ATPase (Gout et al., 1992), which is the major consumer of cellular ATP (Slayman et al., 1973; Sussman, 1994). The inhibition of AOS production by 5  $\mu$ M DPI without the inhibition of the extracellular alkalinization in tobacco cell suspensions treated with cryptogein (Simon-Plas et al., 1997) is another example of the complexity of pH modifications induced by the elicitor. However, taken together, our results indicate that, besides  $\text{H}_2\text{O}_2$  production, the activation of the tobacco plasma membrane redox system by cryptogein appears to be involved in cytoplasmic acidification and extracellular alkalinization.

NAD(P)H oxidoreductase activities associated with plasma membrane of both animal and plant cells have been demon-

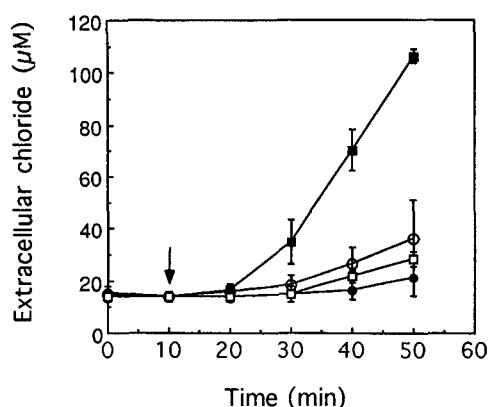


**Figure 7.** Effect of Glucosamine and DPI on the Membrane Depolarization Induced by Cryptogein.

**(A)** Glucosamine (5 mM) was added to the perfusion medium 30 min before cryptogein treatment. Cryptogein (Cr) (150 nM; first arrow) induced membrane depolarization that was reversed (second arrow) by lanthanum ( $\text{La}^{3+}$ ). The bath medium is as given in the legend to Figure 6, except that glucose was replaced by mannitol. Data are representative of five experiments.

**(B)** The addition of 25  $\mu$ M DPI to the perfusion medium (first arrow) was followed 10 min later by a 150-nM cryptogein treatment, which induced a membrane depolarization reversed by lanthanum (third arrow). The same bath medium as given in Figure 6 was used. Data are representative of five experiments.





**Figure 8.** Time Course of Variations in External Chloride Concentration Induced by 50 nM Cryptogein in the Presence or Absence of 250  $\mu$ M Lanthanum.

Cryptogein (50 nM) was added at time 10 min (arrow) in the absence (black squares) or presence (open squares) of 250  $\mu$ M lanthanum added at time 0. Control cells were monitored in the absence (black circles) or presence (open circles) of lanthanum. Each curve is the mean  $\pm$  SE of three independent experiments.

strated (Crane et al., 1988; Chanock et al., 1994). Despite increasing interest, very little is known about the proteins and the possible cofactors of these plant enzymes *in vivo* (Berczi and Asard, 1995).

An NADPH oxidase is present in lymphocytes and in the wall of the endocytic vacuole of phagocytic cells where oxidation of NADPH leads to the reduction of oxygen to superoxide and  $H_2O_2$  (Rossi, 1986). The efficiency of this system resides in the alkalization of vacuole lumen that results from electron transfer unaccompanied by protons and the consumption of protons in the vacuole lumen. This alkalization activates neutral proteinases involved in pathogen digestion (Segal et al., 1981). The components of this redox system and the activation process are now well characterized (reviewed in Segal and Abo, 1993). Thus, in tobacco cells, the redox system activated by cryptogein resembles the NADPH oxidase of phagocytes more than it does the other plant plasma membrane oxidases, considering the electron donor, the electron acceptor, the transient activation, and the involvement in pH alterations. Moreover, tobacco cells contain a protein immunologically related to the neutrophil small G protein Rac2, which is involved in cryptogein-induced oxidative burst (Kieffer et al., 1997).

In animals it has been known for many years that changes in intracellular pH are involved in many processes such as cellular proliferation induced by growth factors (Isfort et al., 1993) or cell death (Greenspan and Aruoma, 1994). Specific inhibitors of vacuolar-type  $H^+$ -ATPase, which provoke cytosol acidification, also induce cell death (Nishihara et al., 1995).

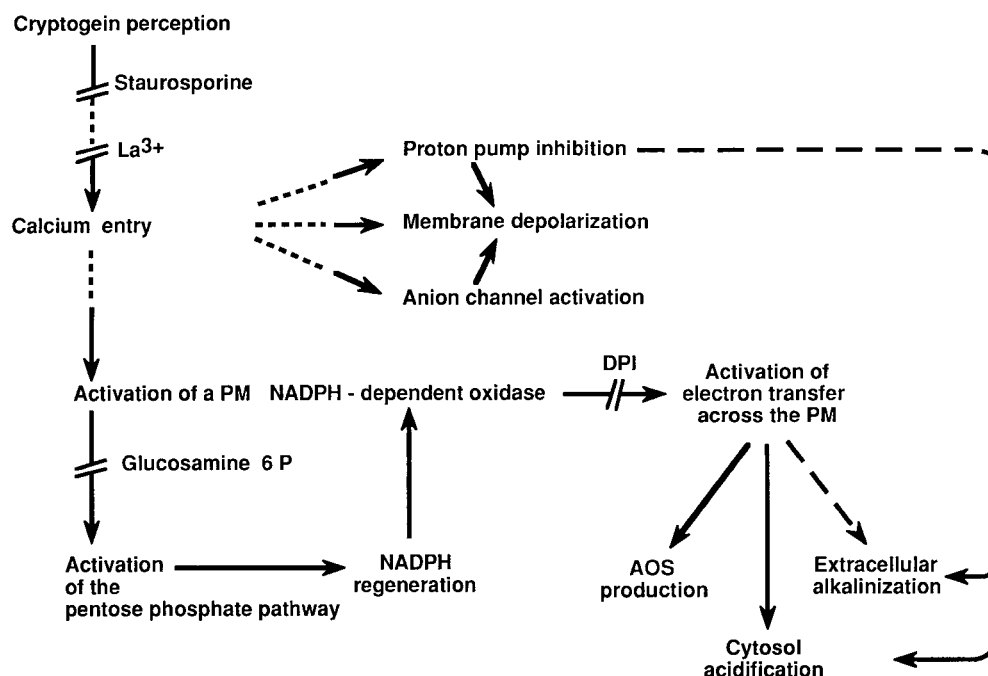
In plant cells, cytosolic pH changes have been shown to be second messengers in signal transduction (Guern et al., 1991). The hypersensitive response to pathogens or elicitors, characterized by the appearance of necrotic lesions, is assumed to be a programmed cell death (Greenberg et al., 1994). Expression of a bacterial proton pump in transgenic tobacco plants activates a cell death program, with the appearance of lesions that resemble the hypersensitive response, and defense reactions (Mittler et al., 1995). Thus, cytoplasm acidification induced by cryptogein in tobacco plants could be a component of the signal cascade that leads to the hypersensitive response.

The activation of a plasma membrane redox system by extracellular ferricyanide has been shown to strongly depolarize cells of different plants (Marré et al., 1988). In neutrophils, activation of the plasma membrane oxidase triggers a membrane depolarization (Henderson et al., 1987). In tobacco cells, the depolarization still occurring when the redox system was inhibited by using DPI or glucosamine-treated cells (Figure 7) indicates that plasma membrane depolarization does not depend directly on the activation of the redox system.

The activation of calcium influx and chloride efflux (Figure 8) as well as an inhibition of the  $H^+$ -ATPase could contribute to the membrane depolarization. The large influx of calcium ( $10 \pm 0.25$  nmol of  $Ca^{2+}$  per g fresh weight of cells per min) corresponds to a depolarizing flux of charges quite similar to that carried by chloride ions ( $28.3 \pm 1.8$  nmol per g fresh weight of cells per min). The inhibition of chloride efflux by lanthanum suggests that the activation of the anion channel depends on the increase in cytoplasmic free calcium and/or on plasma membrane depolarization induced by calcium influx. Many results suggest now that anion channels along with calcium channels play a crucial role in early events of signal transduction (Schroeder and Hagiwara, 1989; Schroeder and Hedrich, 1989; Gilroy et al., 1990; Schroeder et al., 1993; Thuleau et al., 1994; Zimmermann et al., 1994, 1997; Schroeder, 1995; Ward et al., 1995; Gelli et al., 1997).

Figure 9 summarizes our results concerning the organization of the cryptogein responses, as revealed by preventing the massive calcium influx by staurosporine. After cryptogein perception and protein phosphorylation, the influx of calcium triggers two sets of responses: (1) anion efflux and  $H^+$ -ATPase inhibition resulting in plasma membrane depolarization and (2) the activation of a plasma membrane NADPH oxidase responsible for AOS production, cytoplasmic acidification, and extracellular medium alkalization.

In conclusion, many responses of tobacco cells to the elicitor cryptogein have been characterized. Our work contributes to the understanding of the organization of this set of responses. A complex network of putative second messengers, including free calcium, cytosolic and extracellular pH changes, changes in membrane potential, generation of AOS, variations in redox state, and strong metabolic changes, is generated by the elicitor. The role(s) of possible second messengers, individually or in combination, should be more precisely investigated. Plasma membrane proteins, that is,



**Figure 9.** Proposed Model for the Organization of the Early Responses of Tobacco Cells to Cryptogein.

This tentative organization of the elicitor-induced events was built taking into account the effects of inhibitors (staurosporine for protein kinases, lanthanum and EGTA for calcium influx, and DPI, N-Glc-6-P (Glucosamine 6 P), and glycerol 3-phosphate for NADPH oxidase) and not the lag times of the responses, which depended heavily on the conditions of measurement of each response and on the sensitivity of the different methods used. Solid and broken arrows indicate established and hypothetical links, respectively, between events. La<sup>3+</sup>, lanthanum; PM, plasma membrane.

calcium and anion channels, NADPH oxidase, and H<sup>+</sup>-ATPase, which appear as key control proteins of the responses of tobacco cells, are now being investigated.

## METHODS

## Plant Material

Cell suspensions of *Nicotiana tabacum* cv Xanthi were grown in 250-mL conical flasks at 25°C and shaken at 150 rpm under constant light (Milat et al., 1991). The cells came from a culture of leaf tissue. Cell suspensions were subcultured every 7 days and used for elicitor treatment 5 days after subculturing. Cells were also cultivated in the presence of 100 mM glycerol as the sole source of carbon for 3 weeks. The culture medium was changed every 5 days. Cryptogein was purified according to Ricci et al. (1989) and added to cell suspensions as an aqueous solution.

### Extracellular Alkalinization and Active Oxygen Species Production

Elicitor treatments were performed using 1 g fresh weight of cells in 10 mL of 2 mM Mes, 0.5 mM CaCl<sub>2</sub>, 0.5 mM K<sub>2</sub>SO<sub>4</sub>, and 175 mM

mannitol, pH 5.75, previously equilibrated for 2 hr. Extracellular pH was monitored for 1 hr after elicitor addition.

The production of active oxygen species (AOS) by cells was measured by chemiluminescence (luminometer Lumat 9501; Berthold, Bad Wilbad, Germany), using luminol as reagent. After the addition of cryptogein, aliquots of cell suspensions (250  $\mu\text{L}$ ) were added every 10 min to 350  $\mu\text{L}$  of a 10-mM Mes buffer, pH 6.5, containing 175 mM mannitol, 0.5 mM  $\text{CaCl}_2$ , 0.5 mM  $\text{K}_2\text{SO}_4$ , and luminol (final concentration 25  $\mu\text{M}$ ). Chemiluminescence measurements were performed for 10 sec and expressed as  $\text{H}_2\text{O}_2$  equivalents. Assays for monitoring  $\text{H}_2\text{O}_2$  production were performed by adding different known amounts of  $\text{H}_2\text{O}_2$  to cell suspensions, in the range of concentrations measured in experiments with cryptogein-treated cells. The increase in chemiluminescence was linearly proportional to the amount of  $\text{H}_2\text{O}_2$  added between 15 and 150  $\mu\text{M}$ .

### Ca<sup>2+</sup> Influx Measurements

Ca<sup>2+</sup> uptake measurements were performed by the addition of <sup>45</sup>Ca<sup>2+</sup> (0.033 MBq/g fresh weight of cells) 5 min before treatment with cryptogein. After various periods of treatment (0 to 90 min), duplicate samples of 3 mL were withdrawn by filtration and washed once for 1 min and twice for 20 sec on GF/A glass microfiber filters (Whatman) with 10 mL of ice-cold 2 mM LaCl<sub>3</sub> in assay medium without Ca<sup>2+</sup> to remove extracellular <sup>45</sup>Ca<sup>2+</sup>. Cells were scraped from the filters, placed in scintillation vials, and weighed. Ten milliliters of

Ready Safe cocktail (Beckman Instruments, Fullerton, CA) was added to the vials, which were gently shaken overnight before counting in a scintillation counter (model LS 600 TA; Beckman Instruments).

In a second step,  $\text{Ca}^{2+}$  net uptake was measured by flame photometry (AA 300; Varian, Melbourne, Australia). The experiments were performed as described above without  $^{45}\text{Ca}^{2+}$ . The decrease in extracellular  $\text{Ca}^{2+}$  was measured in the assay medium after filtration.

### Chloride Efflux Measurements

To study the efflux of  $\text{Cl}^-$ , a 5-day-old tobacco cell suspension was equilibrated for 5 hr in 10 mM Mes, 5 mM sucrose, 0.5 mM  $\text{CaSO}_4$ , and 3 mM KOH, pH 5.7, at  $\sim 100$  mg fresh weight of cells per mL. After treatment with cryptogein and other effectors, 3.5 mL of the cell suspension was sampled every 10 min and filtered. A 2-mL aliquot of filtrate was adjusted to 0.1 M sulfate with 0.5 M  $\text{K}_2\text{SO}_4$ , and its chloride activity was monitored using a  $\text{Cl}^-$  specific electrode (XS 21 Tacussel; Radiometer, Copenhagen, Denmark) connected with a mercurous sulfate ( $\text{Hg}/\text{Hg}_2\text{SO}_4/\text{K}_2\text{SO}_4$ ) reference electrode to a ION 85 ionometer (Radiometer). The potential of the  $\text{Cl}^-$  specific electrode relative to the reference was linear ( $58 \text{ mV decade}^{-1}$ ) for  $\text{Cl}^-$  concentrations  $>10^{-4}$  M. For each experiment, the electrode was calibrated within a range of  $10^{-6}$  to  $10^{-3}$  M  $\text{Cl}^-$  with KCl. The  $\text{Cl}^-$  concentration of the medium was determined by using the calibration curve.

### In Vitro Phosphorus-31 Nuclear Magnetic Resonance Measurements

#### Perchloric Extract Preparation

Cells (9 g fresh weight) were quickly frozen in liquid nitrogen and ground to a fine powder with a mortar and pestle with 1 mL of 70% (v/v) perchloric acid. The frozen powder was then placed at  $-10^\circ\text{C}$  and thawed. The thick suspension thus obtained was centrifuged at  $15,000g$  for 10 min. The supernatant was neutralized with 2 M  $\text{KHCO}_3$  to approximately pH 5 and then centrifuged at  $10,000g$  for 10 min to remove  $\text{KClO}_4$ . The resulting supernatant was lyophilized and stored in liquid nitrogen. This freeze-dried material was redissolved in 2.5 mL of water containing 10% heavy water ( $\text{D}_2\text{O}$ ) neutralized to pH 7.5 and buffered with 50 mM Hepes. Divalent cations (particularly  $\text{Mn}^{2+}$  and  $\text{Mg}^{2+}$ ) were chelated by the addition of 1,2-cyclohexylenedinitrilotetraacetic acid ranging from 50 to 100 mmol.

### Nuclear Magnetic Resonance Measurements

Phosphorus-31 nuclear magnetic resonance ( $^{31}\text{P}$ -NMR) spectra of neutralized perchloric acid extracts were obtained on an NMR spectrometer (model AM 400, narrow bore; Bruker, Wissenbourg, France) equipped with a 10-mm multinuclear probe tuned at 161.92 MHz. Acquisition used  $70^\circ$  pulses at 3.6-sec intervals. The deuterium resonance of  $\text{D}_2\text{O}$  was used as a lock signal, and the spectra were recorded over a period of 1 hr (1024 scans), or 8 hr when necessary, under conditions of broad-band proton decoupling (Waltz pulse sequence at a sweep width of 6000 Hz). An exponential multiplication (0.2-Hz line width) was used to increase signal-to-noise ratio. Perchloric acid extract spectra are referenced to methylene diphospho-

nic acid, pH 8.9, at 16.38 ppm. The spectra of standard solutions of known phosphate compounds at pH 7.5 were compared with that of a perchloric acid extract of *Acer pseudoplatanus* cells. The definitive assignments were made after running a series of spectra obtained by addition of the authentic compounds to the perchloric acid extracts, according to previous publications (Roby et al., 1987; Aubert et al., 1994).

For the identification and quantification of pyridine nucleotides, perchloric acid-pretreated compounds were used. To determine accurately the total amount of the most abundant soluble organic compounds present in the perchloric extracts, we proceeded as follows: (1) 20-sec recycling time was used to obtain fully relaxed spectra; and (2) a calibration of the peak intensities was measured by integration of peak areas after the addition of known amounts of the corresponding authentic compounds. In addition, we took into account the extraction rate of the perchloric extraction. By adding known amounts of authentic compounds to the frozen cells before grinding, we estimated that the overall yield of recovery was 75 to 80%.

$\text{NADP}^+$  and  $\text{NADPH}$  were measured by  $^{31}\text{P}$ -NMR from the downfield  $\text{NADP}^+$  and  $\text{NADPH}$  singlets at 3.61 and 3.63 plus 3.64 ppm, respectively. It should be stressed that  $\text{NADPH}$  is transformed by perchloric acid extraction in a way that splits and slightly shifts the downfield peak and gives upfield quadruplets between  $-10.6$  and  $-11$  ppm, overlapping with UDP-glucose (UDP-Glc) and  $\alpha$ -ATP peaks, which consequently cannot be used for quantification as shown for mitochondrial extracts (Roberts et al., 1997). In contrast, oxidized pyridine nucleotides ( $\text{NADP}^+$  and  $\text{NAD}^+$ ) are not modified by perchloric acid extraction, giving an upfield quadruplet at  $-10.87$ ,  $-10.99$ ,  $-11.19$ , and  $-11.31$  ppm.  $\text{NAD}^+$  was measured from the upfield  $\text{NAD}^+$  plus  $\text{NADP}^+$  quartet after subtraction of the contribution of  $\text{NADP}^+$  calculated from the downfield peak of  $\text{NADP}^+$  (at 3.61 ppm). Concentrations were estimated assuming that these compounds are present exclusively in the cytoplasm and that the cytoplasm occupies nearly 10% of the total cell volume, as described by Aubert et al. (1994).

### In Vivo $^{31}\text{P}$ -NMR Measurements

Ten grams fresh weight of tobacco cells was slightly compressed in a 25-mm NMR tube and perfused under slight pressure at a flow rate of 50 mL per min with a well-oxygenated ( $\text{O}_2$  bubbling) nutrient medium (0.5 mM KCl, 0.5 mM  $\text{KNO}_3$ , 0.5 mM  $\text{MgSO}_4$ , 1 mM  $\text{Ca}(\text{NO}_3)_2$ , 0.1 mM  $\text{KH}_2\text{PO}_4$ , and 2 mM sucrose, pH 6.3). After 2 hr of equilibration, 100 nM cryptogein was added in this circulating medium. The pH of this medium (500 mL) was maintained constant by using a pH stat coupled to a titrimeter monitoring the addition of registered amounts of HCl or KOH.

To get a better signal-to-noise ratio, an experimental arrangement was realized to analyze the maximum cell volume and to optimize the homogeneity of the cell incubation conditions (Roby et al., 1987).  $^{31}\text{P}$ -NMR spectra were measured on a spectrometer (AMX 400, wide bore; Bruker) equipped with a 25-mm probe tuned at 161.92 MHz. Acquisition used  $50^\circ$  pulses at 0.6-sec intervals, and free induction decays were recorded with 8000 data points and zero filled to 16,000 before Fourier transformation. A 20-sec recycling time was used to obtain fully relaxed spectra necessary for quantitative measurements. The deuterium resonance of  $\text{D}_2\text{O}$  was used as a lock signal, and the spectra were recorded over a period of 10 min (1000 scans) under conditions of broad-band proton decoupling (Waltz pulse sequence) at a sweep width of 6000 Hz. An exponential multiplication (5-Hz line width) was used to increase the signal-to-noise ratio. Spectra are referenced to a solution of 50 mM methylene diphosphonic

acid (pH 8.9 in 30 mM Tris) contained in a 0.8-mm capillary inserted inside the inlet tube along the symmetry axis of the cell sample (Roby et al., 1987).

The intracellular pH values were first estimated from the intracellular Pi pools (cytoplasm and vacuole), as indicated by Gout et al. (1992). In addition, methylphosphonate added at a concentration of 1 mM for 6 hr in the culture medium was also used as a convenient probe for the precise determination of cytoplasmic pH. Indeed, methylphosphonate has a chemical shift sensitivity to pH nearly twice that of Pi, a pKa of 7.6, and it resonates between 20 and 25 ppm in a region usually devoid of other phosphorus-31 resonances. For this reason, it has already been used successfully with other materials (Slonczewski et al., 1981; Satre et al., 1989).

The assignment of Pi, phosphate esters, phosphate diesters, and nucleotides to specific peaks was performed according to the methods of Roberts and Jardetzky (1981), Roby et al. (1987), and Aubert et al. (1994) and from spectra of the perchloric acid extracts that contained the soluble low molecular weight constituents. Finally, when given, intracellular concentrations were calculated on the following basis: 1 g cell wet weight corresponds to 1 mL cell volume and ~0.10 mL cytoplasm.

### Measurement of Membrane Potentials

A 5-day-old tobacco cell suspension was equilibrated during 3 hr in 10 mM Mes, 5 mM glucose, 0.5 mM CaSO<sub>4</sub>, and 3 mM KOH, pH 5.7. To immobilize cells, the cut section of a 3-mm-long capillary was covered with a thin layer of lanolin and then dipped into the cell suspension. The capillary with immobilized cells at its section was fixed at the bottom of a perfusion chamber.

Micropipettes were pulled with a vertical puller (model PA-81; Narishige, Tokyo, Japan) from filament containing borosilicate glass (model GC 150F-15; Clark, Pangbourne, UK). The reference electrode (positioned downstream in the perfusion chamber) and the micropipette were filled with 0.5 M KCl. The resistance of the microelectrode was ~30.10<sup>6</sup> ohms. The electrodes were connected via Ag/AgCl wires to a high-impedance electrometer amplifier (model 8700 cell explorer; Dagan, Minneapolis, MN). The voltage changes were recorded with a pen chart recorder.

The chamber (2-mL volume) was continuously perfused at a rate of 2.5 mL per min. After impalement, when a stable membrane potential was obtained for several minutes, the bath solution was supplemented with the different effectors to be tested.

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